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Key indicators

Single-crystal X-ray study T = 296 KMean σ (C–C) = 0.007 Å R factor = 0.057 wR factor = 0.158 Data-to-parameter ratio = 10.1

For details of how these key indicators were automatically derived from the article, see http://iournals.jucr.org/e.

3β-Hydroxyolean-12-en-27-oic acid methanol solvate: a cytotoxic and apoptosis-inducing triterpenoid from the rhizome of Astilbe chinensis

The title compound, C₃₀H₄₈O₃·CH₄O, is a cytotoxic and apoptosis-inducing triterpenoid which was isolated from the rhizome of Astilbe chinensis. The molecule is composed of five six-membered rings. The cyclohexene ring adopts a slightly distorted half-chair conformation and the cyclohexane rings adopt chair conformations. The hydroxy and carboxy groups both serve simultaneously as hydrogen-bond donors and acceptors, forming molecular chains.

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Comment

Astilbe chinensis (Maxim.) Franch. et Savat. (Saxiffagaceae) is a perennial herbaceous plant growing at an altitude of 390-3600 m in China, Russia, Japan and Korea. Its rhizome, known as 'Luo Xinfu' (Chinese name), has been used for headache, arthralgia, chronic bronchitis and stomachalgia, in traditional Chinese medicine (Pan, 1985, 1995). Pharmacological experiments have indicated that the extracts from A. chinensis have antineoplastic and immunopotent activities (Chen et al., 1996). Previously, we isolated β -sitosterol palmitate, daucosterol, β -sitosterol and bergenin from the rhizome of A. chinensis (Sun et al., 2002). To investigate the bioactive natural products from A. chinensis, chemical studies on the rhizome of A. chinensis were undertaken by screening using antineoplastic tests in vitro, and we have obtained cytotoxic and apoptosisinducing triterpenoid 3β -hydroxyolean-12-en-27-oic acid (I), from its petroleum ether extract. It exhibits significant cytotoxic activity against human ovarian carcinoma cell line (HO-8910), human cervical squamous carcinoma cell line (HeLa) and human leukemia cell (HL60) in vitro; its 50% inhibitory concentration (IC₂₀50) values were 8.0(9), 3.94(13) and 3.67 (19) μ g ml⁻¹, respectively. Moreover, it induces apoptotic cell death of the HO-8910 in a dose-dependent manner, ranging from 2.5 μ g ml⁻¹ to 40.0 μ g ml⁻¹. The structure of the title compound, (I), was elucidated by spectroscopic analysis including two-dimensional NMR spectroscopy and confirmed by single-crystal X-ray diffraction analysis.



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The structure of (I) with the atom-numbering scheme is shown in Fig. 1. The molecule is composed of five six-

organic papers

organic papers



Figure 1

A view of (I), with the atomic numbering scheme. Displacement ellipsoids are drawn at the 35% probability level. H atoms have been omitted for clarity.

membered rings, *viz*. A (C1–C5/C10), B (C5–C10), C (C8/C9/C11–C14), D (C13–C18) and E (C17–C22). Rings A, B, D and E adopt chair conformations, while ring C adopts a slightly distorted half-chair conformation owing to the C12=C13 double bond. All rings are *trans* fused except for the D/E junction, which is *cis* fused. The orientation of the carboxy group is axial and that of the hydroxy group is equatorial.

The hydroxy and carboxy groups both serve simultaneously as hydrogen-bond donors and acceptors (Table 2). In the asymmetric unit, the two molecules are linked by O2– $H2\cdots O4$ hydrogen bonds. In the solid state, the screw-related molecules are linked by O1– $H1\cdots O3^{i}$ hydrogen bonds (Table 2), to form molecular chains along the *a* axis. The chain formation is further stabilized by the methanol solvent through O4– $H4O\cdots O1^{i}$ hydrogen bonds. The crystal structure is further stabilized by C– $H\cdots O$ interactions.

Experimental

The rhizomes of Astilbe chinensis were collected in Anji county, Zhejiang province, China in August 2001. The plants were identified as A. chinensis (Maxim.) Franch. et Savat. by Professor Xiang-Ji Xue, College of Pharmaceutical Science, Zhejiang University. A voucher specimen (No. 200120) was deposited in the Laboratory of Nature and Biochemistry, Zhejiang University. The rhizomes of Astilbe Chinensis were dried at 313 K in the dark, in a ventilated hood, and ground. The material (5.0 kg) was extracted three times with CH₃OH (each 25 l) at room temperature. The CH₃OH extract (366 g) was partitioned between petroleum ether (201) and water (21). The petroleum ether solution was concentrated in vacuum to give 41.2 g of a gelatinous material. A part of the extract (40.0 g) was absorbed on to silica gel (60 g) and chromatographed on a silica gel (600 g) column eluted with petroleum ether-EtOAc (50:1, 30:1, 15:1, 5:1, 3:1, 2:1) gradients. The eluted fractions were evaluated by TLC and combined to give fractions 1 to 19, respectively. Fraction 12 was recrystallized from EtOAc-CH₃OH (1:1) to afford 2.5 g of the pure title compound, (I). m.p.: 513.5–515.5 K. ¹³C NMR (125 MHz, C₅ND₅, p.p.m): 179.4 (C27), 138.2 (C13), 126.2 (C12), 79.4 (C3), 55.8 (C14), 55.2 (C5), 49.0 (C18), 47.2 (C9), 43.9 (C19), 39.7 (C8), 7 (C4), 38.6 (C1), 37.1 (C10), 36.5 (C21), 36.3 (C7), 34.3 (C22), 33.5 (C29), 38 32.8 (C17), 31.1 (C20), 28.2 (C28), 28.1 (C23), 27.5(C2), 26.9 (C16), 23.6 (C30), 22.8 (C11), 22.2 (C15), 18.2 (C6), 18.0 (C26), 16.4 (C25), 16.0 (C24). Crystals suitable for X-ray structure analysis were obtained by slow evaporation from methanol at room temperature.

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\begin{array}{l} C_{30}H_{48}O_{3}{\cdot}CH_{4}O\\ M_{r}=488.73\\ Orthorhombic, P2_{1}2_{1}2_{1}\\ a=7.538\ (2)\ \text{\AA}\\ b=13.710\ (3)\ \text{\AA}\\ c=27.629\ (8)\ \text{\AA}\\ V=2855.3\ (13)\ \text{\AA}^{3}\\ Z=4\\ D_{x}=1.137\ \text{Mg}\ \text{m}^{-3} \end{array}
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Data collection

Siemens *P*4 diffractometer ω scans Absorption correction: none 3492 measured reflections 3203 independent reflections 1876 reflections with *I* > 2 $\sigma(I)$ *R*_{int} = 0.021

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0951P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.057$	where $P = (F_o^2 + 2F_c^2)/3$
$vR(F^2) = 0.158$	$(\Delta/\sigma)_{\rm max} < 0.001$
S = 0.89	$\Delta \rho_{\rm max} = 0.21 \ {\rm e} \ {\rm \AA}^{-3}$
3203 reflections	$\Delta \rho_{\rm min} = -0.19 \mathrm{e} \mathrm{\AA}^{-3}$
316 parameters	Extinction correction: SHELXTL
H-atom parameters constrained	Extinction coefficient: 0.0116 (17)

Mo $K\alpha$ radiation

reflections

 $\theta = 2.9 - 13.8^{\circ}$ $\mu = 0.07 \text{ mm}^{-1}$

T = 296 (2) K

 $\theta_{\rm max} = 26.0^{\circ}$

 $h = 0 \rightarrow 9$

 $k = 0 \rightarrow 16$

 $l = -1 \rightarrow 34$

3 standard reflections

every 97 reflections

intensity decay: 1.5%

Prism, colorless $0.58 \times 0.40 \times 0.24$ mm

Cell parameters from 38

Table 1

Selected geometric parameters (Å, °).

O1-C3	1.449 (5)	C12-C13	1.336 (6)
O2-C27	1.323 (5)	O4-C31	1.369 (8)
O3-C27	1.213 (5)		. ,
O1-C3-C2	107.5 (4)	C12-C13-C14	119.9 (4)
O1-C3-C4	112.4 (4)	C18-C13-C14	120.2 (4)
C6-C5-C4	115.8 (4)	C13-C14-C27	105.2 (4)
C4-C5-C10	116.9 (3)	C16-C15-C14	115.1 (4)
C8-C9-C10	116.3 (3)	O3-C27-O2	121.4 (4)
C13-C12-C11	126.7 (4)	O3-C27-C14	124.6 (4)
C12-C13-C18	119.9 (4)	O2-C27-C14	114.0 (4)
C1-C2-C3-O1	-178.2 (4)	C15-C14-C27-O3	-159.3 (4)
O1-C3-C4-C24	67.2 (5)	C8-C14-C27-O3	78.1 (5)
O1-C3-C4-C23	-49.5(5)	C13-C14-C27-O2	139.9 (4)
O1-C3-C4-C5	-175.7(3)	C15-C14-C27-O2	22.5 (5)
C13-C14-C27-O3	-42.0 (6)	C8-C14-C27-O2	-100.0(4)

Table 2		
Hydrogen-bonding geometry	(Å	°)

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D{\cdots}A$	$D - H \cdot \cdot \cdot A$
O2−H2···O4	0.82	1.82	2.619 (5)	164
$O1-H1\cdots O3^i$	0.82	2.17	2.991 (5)	178
$O4-H4O\cdots O1^i$	0.82	1.96	2.755 (5)	164

Symmetry code: (i) $x - \frac{1}{2}, \frac{1}{2} - y, 1 - z$.

After their location in a difference map, all H atoms were geometrically fixed and allowed to ride on their attached atoms using *SHELXL*97 (Sheldrick, 1997) defaults. The absolute configuration could not be established because of the absence of significant anomalous effects. Friedel pairs were merged for the final cycles of refinement.

Data collection: *XSCANS* (Siemens, 1994); cell refinement: *XSCANS*; data reduction: *XSCANS*; program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *SHELXTL/PC* (Siemens, 1991); software used to prepare material for publication: *SHELXL*97.

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